CHEM 1094: CHEMICAL AND PHYSICAL CHANGES

Date: Name:	Lab Day/Time:
Objective	
To recognize when a chemical reaction has occurred and to make deductions based upon	on careful observations.
Procedure	
As in the Chem 1094 lab manual, pages	
Observations and Data	

Test	Solutions	Observations	Deductions (choose one from each column below)		
1	Calcium chloride and Sodium carbonate		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐	
2	Silver nitrate and Copper		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐	
3	6 M Sodium hydroxide and 6 M Hydrochloric acid		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐	
4	Hydrogen peroxide and Potassium iodide		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐	

Test	Solutions	Observations	Deductions (choose one from each column below)	
5	Ferric chloride and Ammonium thiocyanate		☐ Chemical change ☐ Physical change ☐ Neither	☐ Reaction☐ No reaction☐
6	Step 5 and Silver nitrate		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
7	Potassium permanganate and water		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
8	Potassium permanganate and oxalic acid		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
9	Potassium permanganate and oxalic acid and heat		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
10	Sucrose and Heat		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
11	Sodium Chloride and Heat		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
12	Tin and Heat		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐
13	Ice and Heat		☐ Chemical change☐ Physical change☐ Neither	☐ Reaction☐ No reaction☐

Conclusion

Question

Answer any assigned questions.