KWANTLEN UNIVERSITY COLLEGE CHEMISTRY 1110 S-10 EXAM No. 1 Thursday February 19, 1998

ANSWER KEY

Note: Detailed solutions to the problems are available at the reserve desk in the library.

Question One:

- **a**) 89 s
- **b**) 2.50 *M*
- **c)** 0.01500 N; $7.500 \times 10^{-3} M$

Question Two:

- **a**) 32.9 g/mol
- **b**) **i**) $131g \text{ NaN}_3(s)$
 - **ii)** 94.7%

Question Three: 43.9% Ag

Question Four: Only one of many possible answers is provided for each case below.

c) H
$$C - C$$
 H $C - C$ H

e)
$$H_{3}C-C$$

$$H_{3}C-C$$

$$0$$

$$H_{3}C-C$$

$$0$$

Question Five:

a)

$$^{\text{H}_3\text{C}}$$
 $^{\text{CH}_3}$ $^{\text{CH}_2\text{CH}_3}$

$$\begin{array}{c} O \\ \parallel \\ C \\ \downarrow \\ H_2C \\ C \\ CH_2 \\ H_2C - CH_2 \end{array}$$

d)

Question Six: (10 MARKS)

- a) 3-chloro-4-methyl-pentanol
- **b**) *trans*-3,6-dichloro-1,2-dimethylcyclohexene
- **c)** *m*-bromoethylbenzene or bromo-3-ethylbenzene
- **d**) 3,5-dimethyl-3-hexanol
- e) 3,3-dimethylbutanoic acid

Question Seven:

a)
$$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \\ \text{CH}_3 \end{array}$$
 Br
$$\begin{array}{c} \text{Br} \\ \text{Minor Prod.} \end{array}$$
 Major Prod.

f)
$$O$$
 CH_3 $+ H_2O$

Question Eight:

Examples of the various isomers are presented below. Others are possible.

a) Br O O
$$H_2C=C-CH_2CH_2CH_2CH_3$$
 $CH_3CH_2CH_2CH_2CH_3$ $CH_3CH_2CH_2CH_3$ $CH_3CH_2CH_2CH_3$

Question Nine:

a)
$$HO$$
 CH_3 OH CH_3 CH_3 CH_3 CH_3

Note: In this problem B & C are interchangable.

b)
$$CH_3$$
 CH_3 III CH_3 III IV CH_3 V VI