

**KWANTLEN UNIVERSITY COLLEGE**  
**CHEMISTRY 1110 S-10**  
**EXAM No. 1**  
**Thursday February 19, 1998**

**ANSWER KEY**

**Note:** Detailed solutions to the problems are available at the reserve desk in the library.

**Question One:**

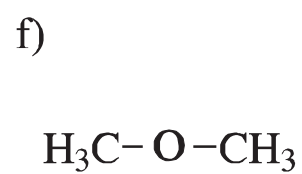
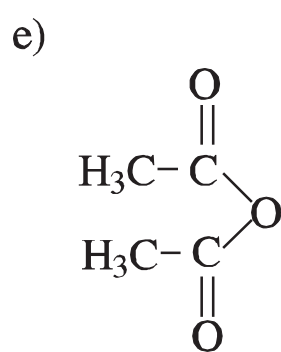
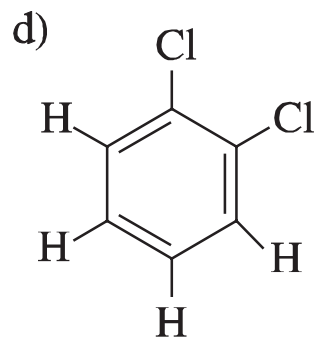
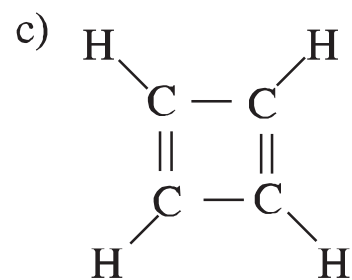
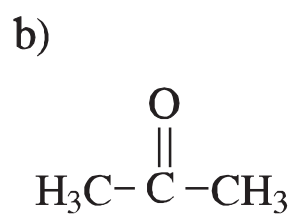
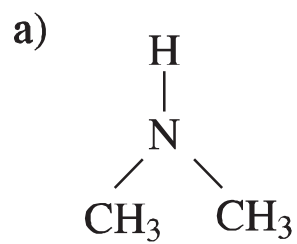
- a) 89 s
- b) 2.50 *M*
- c) 0.01500 *N*;  $7.500 \times 10^{-3}$  *M*

**Question Two:**

- a) 32.9 g/mol
- b) i) 131g NaN<sub>3</sub>(s)  
ii) 94.7%

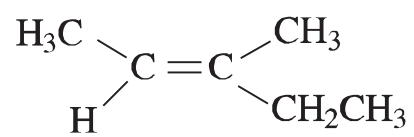
**Question Three:** 43.9% Ag

**Question Four:** Only one of many possible answers is provided for each case below.

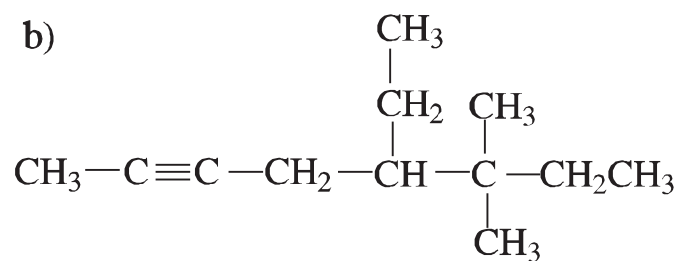


Question Five:

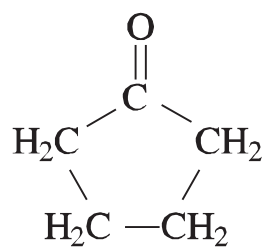
a)



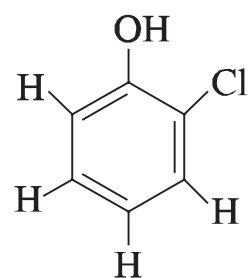
b)



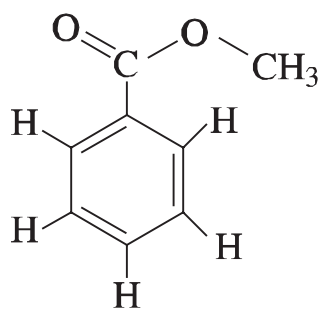
c)



d)



e)

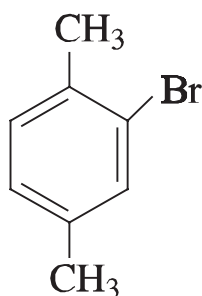


**Question Six: (10 MARKS)**

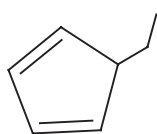
- a) 3-chloro-4-methyl-pentanol                      b) *trans*-3,6-dichloro-1,2-dimethylcyclohexene
- c) *m*-bromoethylbenzene or bromo-3-ethylbenzene                      d) 3,5-dimethyl-3-hexanol
- e) 3,3-dimethylbutanoic acid

**Question Seven:**

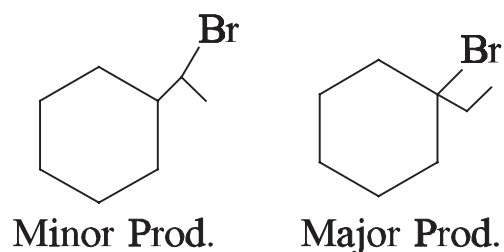
a)



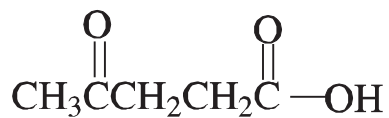
b)



c)



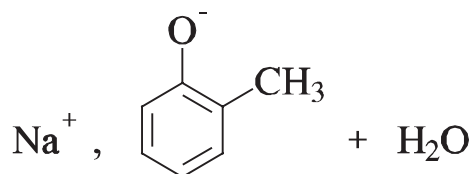
d)



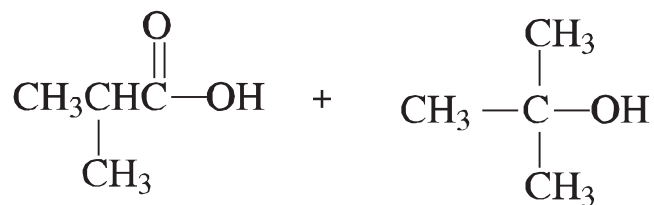
e)



f)

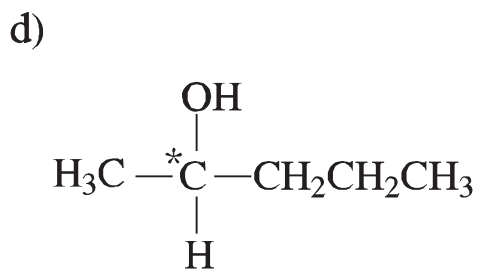
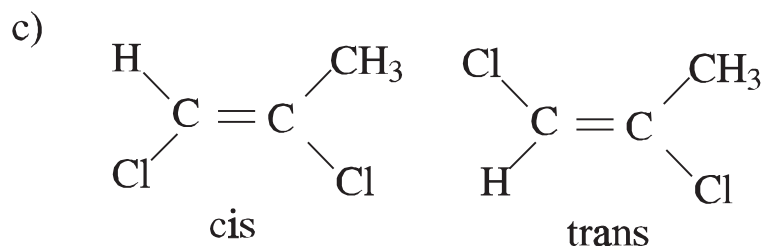
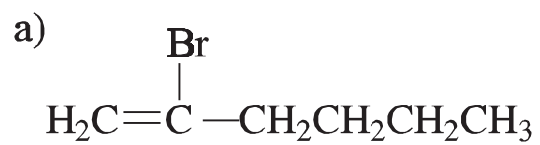


g)

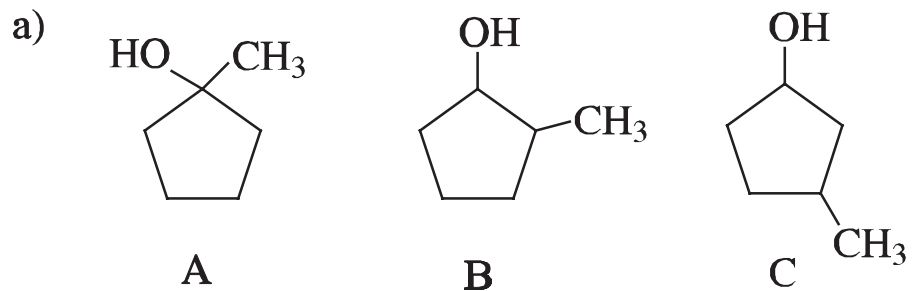


**Question Eight:**

Examples of the various isomers are presented below. Others are possible.



**Question Nine:**



Note: In this problem **B & C** are interchangeable.

